

Quantum dots: tiny materials whose properties are determined by their size

**Lecture Series on Nobel Prizes 2023**

January, 2024

Presented by

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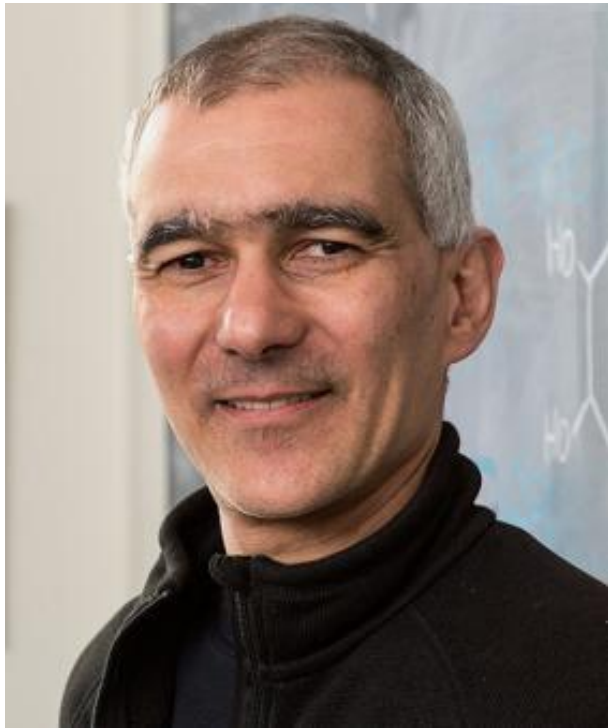
Email: [khatuask@iitgn.ac.in](mailto:khatuask@iitgn.ac.in)

Website: <https://khatuask.wixsite.com/nprl>



# The Nobel Prize in Chemistry 2023

- The Nobel Prize in Chemistry 2023 was awarded to Mounqi G. Bawendi, Louis E. Brus and Aleksey Yekimov "for the discovery and synthesis of quantum dots"



Prof. Mounqi Bawendi  
Massachusetts Institute of  
Technology



Prof. Louis E. Brus  
Columbia University



Prof. Alexei Ekimov  
Vavilov State Optical Institute

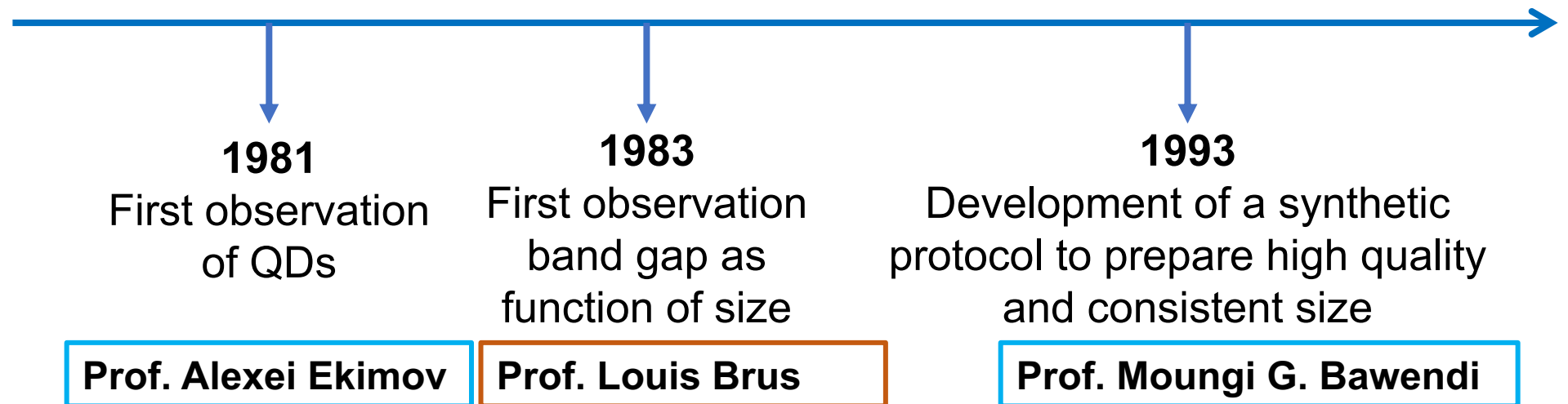
# Presentation Outline

## □ Introduction to QDs

- What are QDs and why are they so special
- Basic theoretical background

## □ Discovery and synthesis of QDs

- A chronological description of the challenges and milestones



## □ Recent developments, challenges and future perspective

- Applications in QLEDs and imaging
- Single photon source and blinking

# Introduction to QDs

❖ Typically, properties of matter are defined by the atoms they are made of

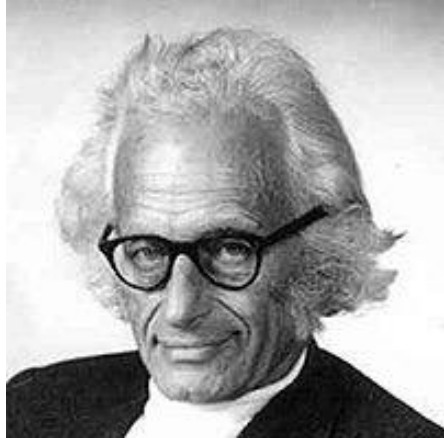


Green dye

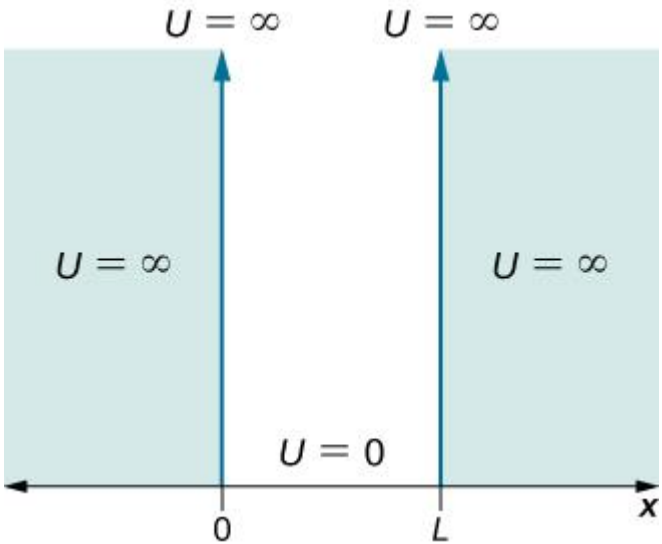
Red dye

# Introduction to QDs

1937: Herbert Froehlich predicted that once particles were small enough—so-called nanoparticles—they would come under the strange spell of quantum mechanics



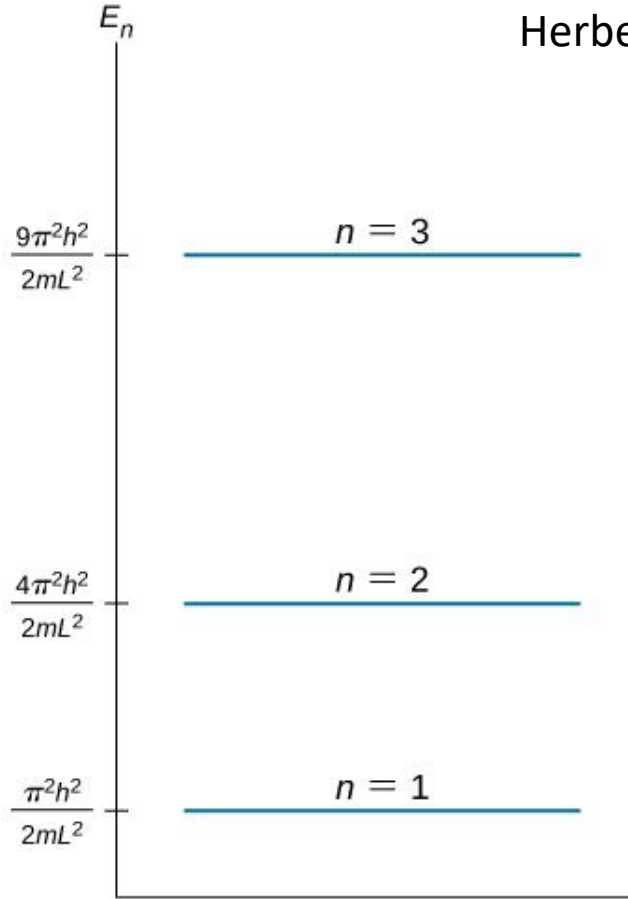
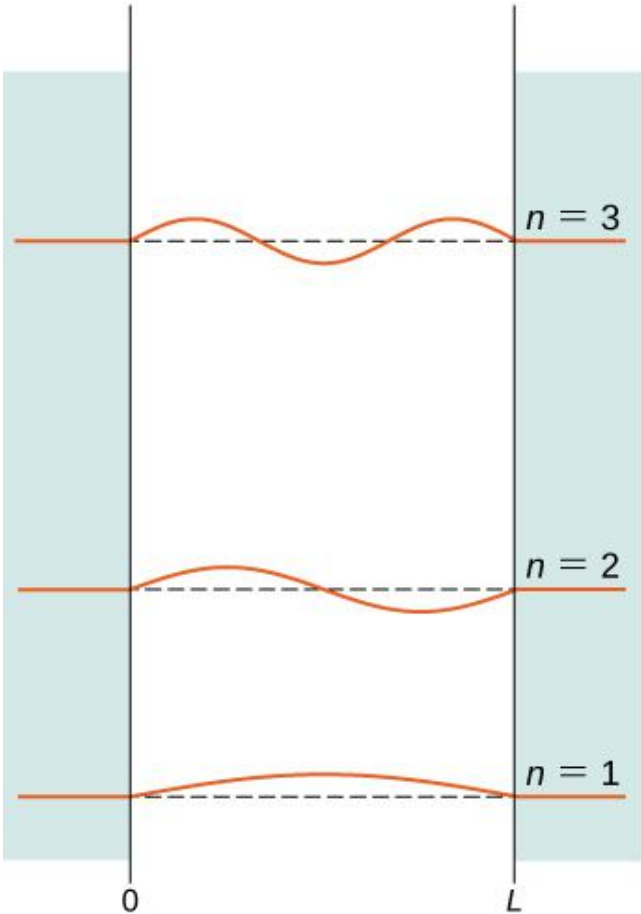
Herbert Fröhlich



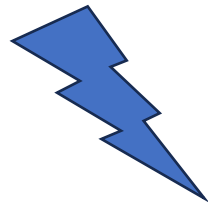
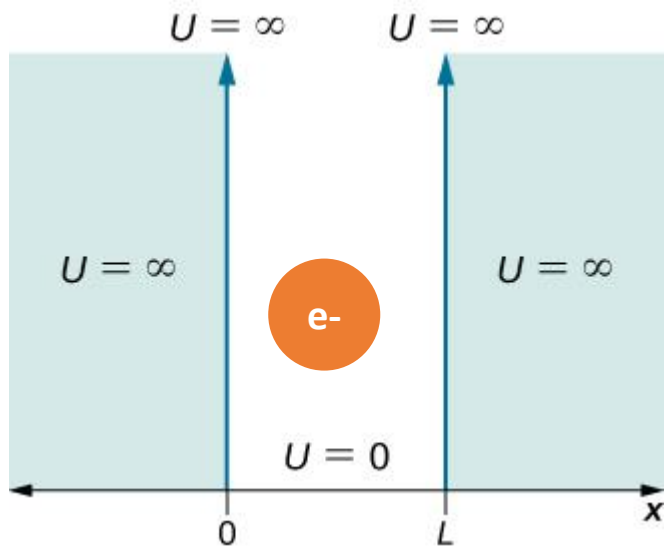
Particle-in-box problem

$$E_n = \frac{n^2 h^2}{8mL^2} \quad \psi_n = A \sin \frac{n\pi x}{L}$$

$n=1,2,3,\dots$



# Introduction to QDs



## Let's assume a scenario

Particle= electron  
 $L=0.6$  nm

$$\Delta E_{1 \rightarrow 2} = \frac{3h^2}{8mL^2} \sim 5 \times 10^{-19} \text{ J}$$

$$\lambda_{1 \rightarrow 2} \sim 400 \text{ nm}$$

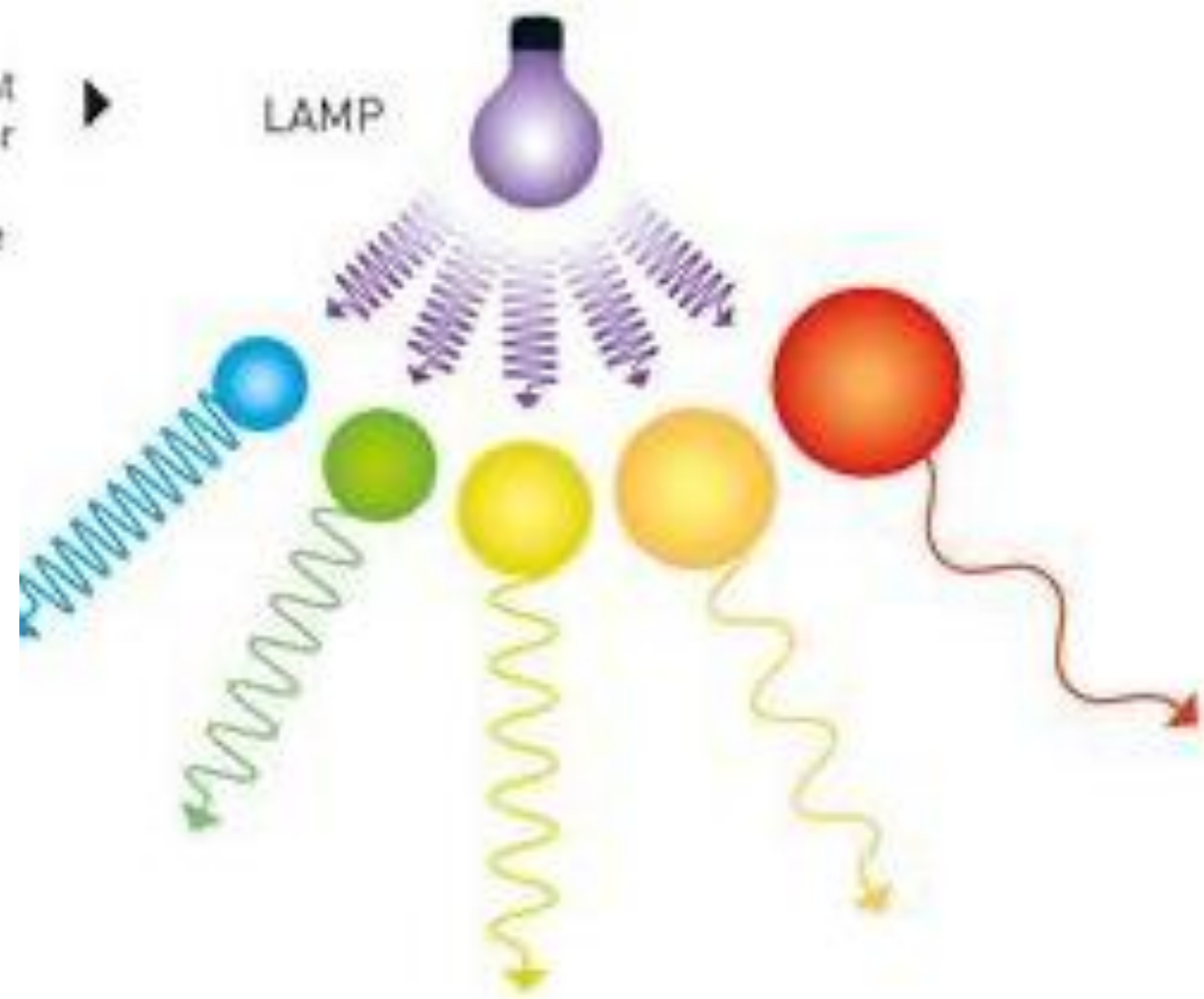
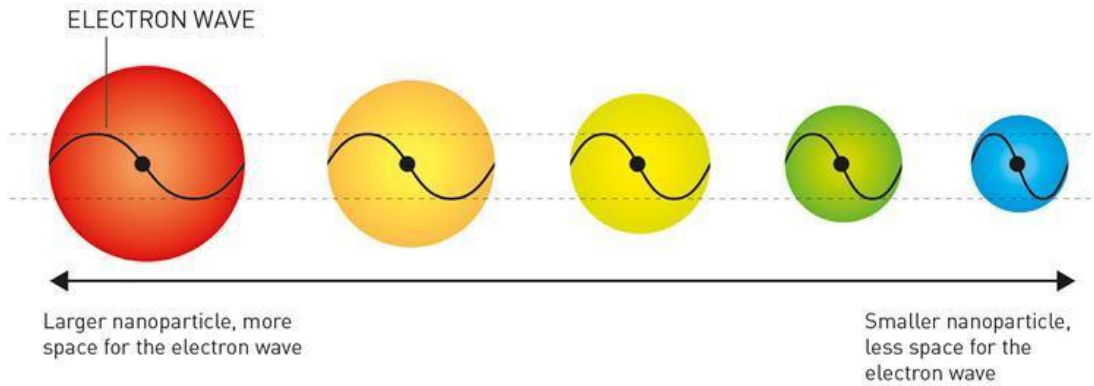
$$\lambda_{n \rightarrow n+1} \propto L^2$$



# Introduction to QDs

Quantum dots absorb light and then emit it at another wavelength. Its colour depends on the size of the particle.

When particles are just a few nanometres in diameter, the space available to electrons shrinks. This affects the particle's optical properties.



# The Journey to realize the size-dependent quantum effect

- Prof. Alexei Ekimov and Prof. Alexander Efros was studying origin of colors in Schott glasses
- They were heating glasses up to 700°C at varying rate and then studied the tiny semiconductor particles formed using X-rays.
- They found formation of tiny copper chloride crystals between 2 nm to 20 nm and they **attributed such size difference to the different color of glasses.**

Published in Russian language in Soviet Scientific Journal in 1981: Remained unnoticed..

The particles were frozen inside the glass so can't be used for any purpose.





# A simple model for the ionization potential, electron affinity, and aqueous redox potentials of small semiconductor crystallites

➤ Prof. Lou Raman S

L. E. Brus



➤ 'I noticed made the the colloidal band gap

*J. Chem. Phys.* 79, 5566–5571 (1983)

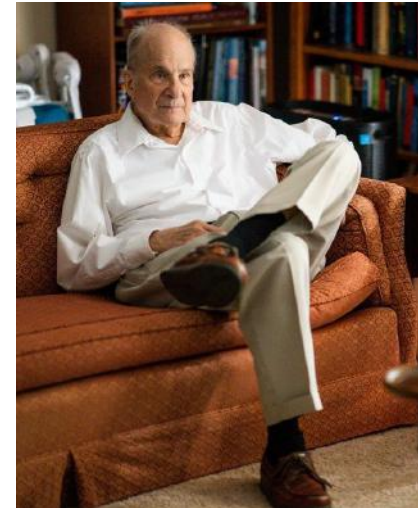
<https://doi.org/10.1063/1.445676> Article history 



➤ 'Solid state important small molecules understood but it was

Large semiconductor crystals have intrinsic electronic properties dependent upon the bulk band structure. As the crystal becomes small, a new regime is entered in which the electronic properties (excited states, ionization potential, electron affinity) should be strongly dependent upon the electron and hole in a confined space. We address the possibility of a shift in the photochemical redox potential of one carrier, as a function of crystallite size. As a semiquantitative guide, one might expect a shift on the order of  $h^2/8em^*R^2$  due to the kinetic energy of localization in the small crystallite. We model the elementary quantum mechanics of a charged crystallite using (a) the effective mass approximation, (b) an electrostatic potential for dielectric polarization, and (c) penetration of the carrier outside the crystallite in a cases of small effective mass. Shifts of several tenths of an eV appear possible in crystallites of diameter 50 Å. The carrier charge density reside near the crystallite surface if the effective mass is very small.

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# The journey to realize the size-dependent quantum effect

- ❖ Brus and Ekimov established the quantum size effects in colloidal nanocrystals but there was still a problem to be solved.

Their methods created quantum dots of unpredictable quality and size, often with defects; a reliable and consistent synthesis method was needed.

‘We struggled to get this colloidal synthesis under control because it was a kind of a black magic in the beginning; it was just a recipe’-  
Louis Brus



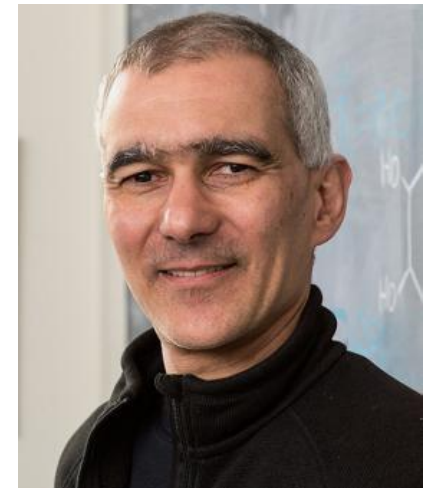
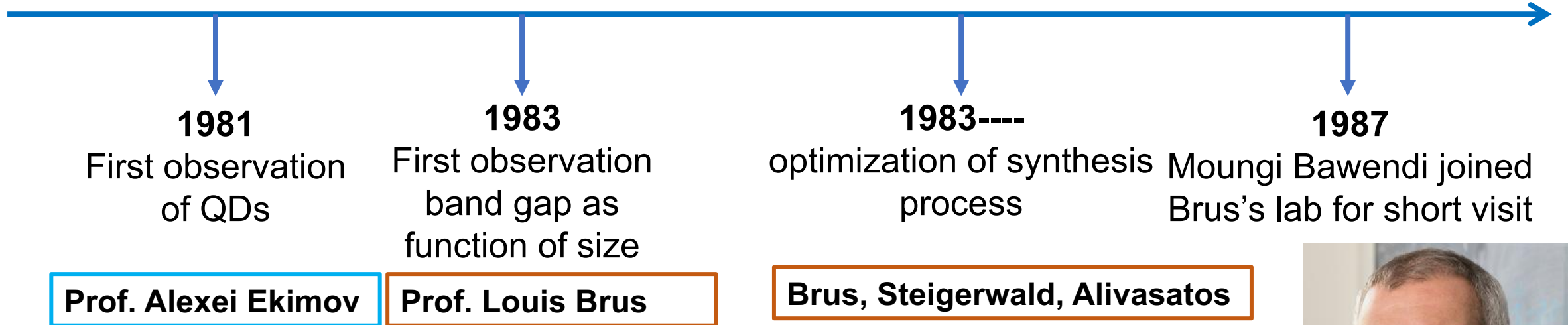
Paul Alivisatos



Michael L. Steigerwald

# The journey to realize size-dependent quantum effect

➤ A chronological description of the challenges and milestones



# The journey to realize size-dependent quantum effect: the synthesis of monodispersed QDs with tunable properties

8706

*J. Am. Chem. Soc.* **1993**, *115*, 8706–8715

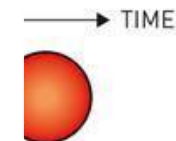
## Synthesis and Characterization of Nearly Monodisperse CdE (E = S, Se, Te) Semiconductor Nanocrystallites

C. B. Murray, D. J. Norris, and M. G. Bawendi\*

*Contribution from the Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139*

*Received March 22, 1993*

**Abstract:** A simple route to the production of high-quality CdE (E = S, Se, Te) semiconductor nanocrystallites is presented. Crystallites from  $\sim 12$  Å to  $\sim 115$  Å in diameter with consistent crystal structure, surface derivatization, and a high degree of monodispersity are prepared in a single reaction. The synthesis is based on the pyrolysis of organometallic reagents by injection into a hot coordinating solvent. This provides temporally discrete nucleation and permits controlled growth of macroscopic quantities of nanocrystallites. Size selective precipitation of crystallites from portions of the growth solution isolates samples with narrow size distributions ( $<5\%$  rms in diameter). High sample quality results in sharp absorption features and strong “band-edge” emission which is tunable with particle size and choice of material. Transmission electron microscopy and X-ray powder diffraction in combination with computer simulations indicate the presence of bulk structural properties in crystallites as small as 20 Å in diameter.



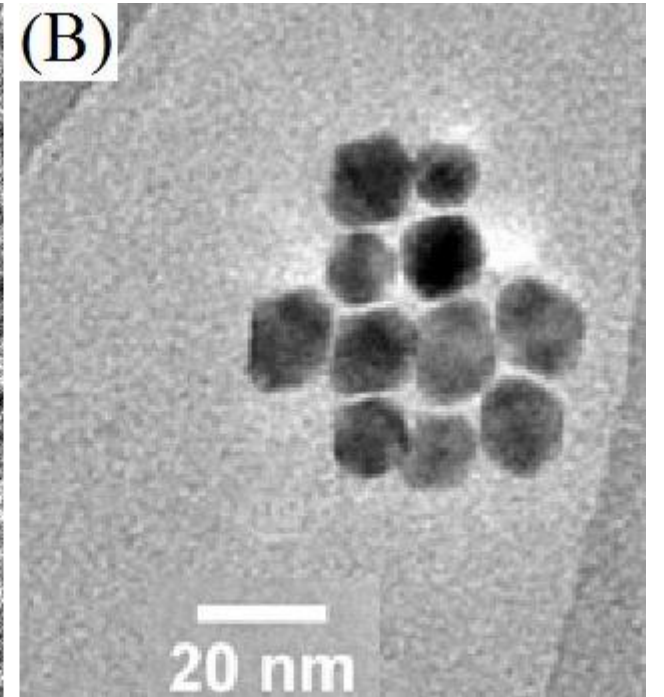
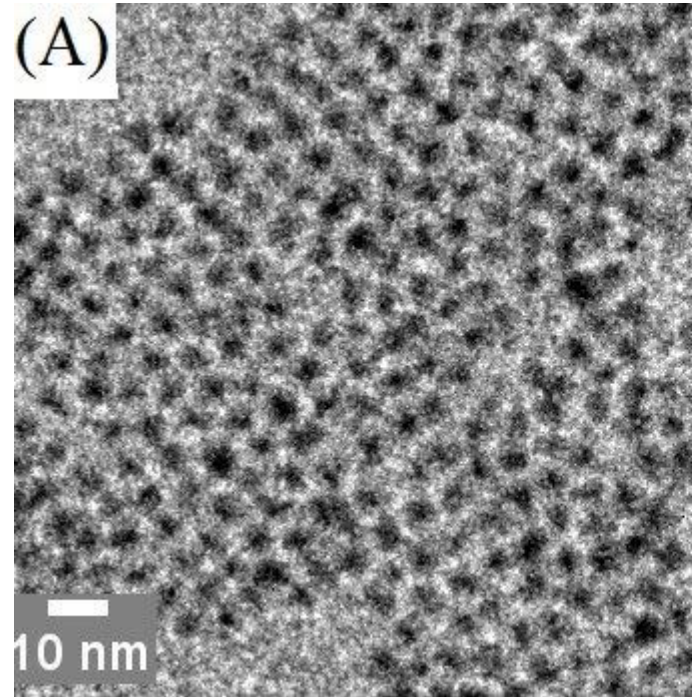
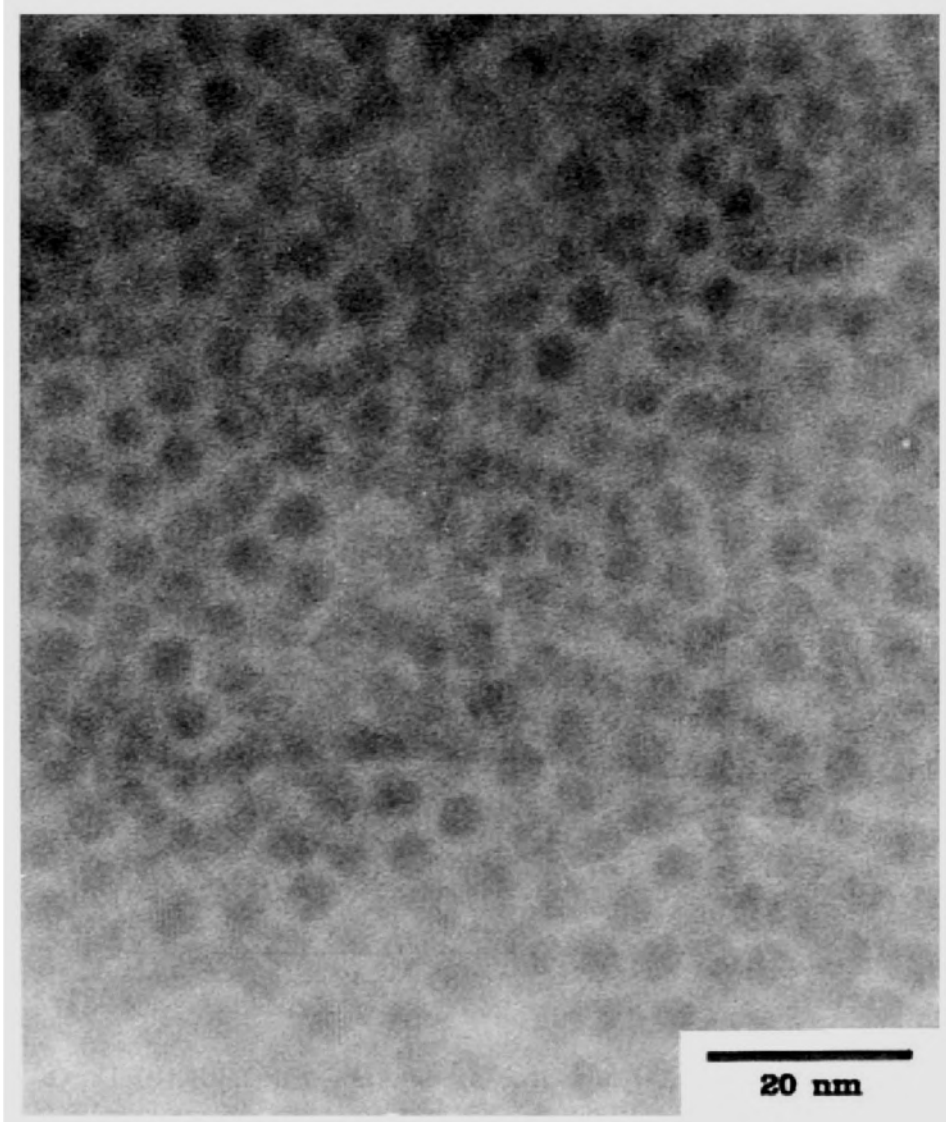
Chris Murray

David Norris

Bawendi



# Synthesis of Quantum Dots





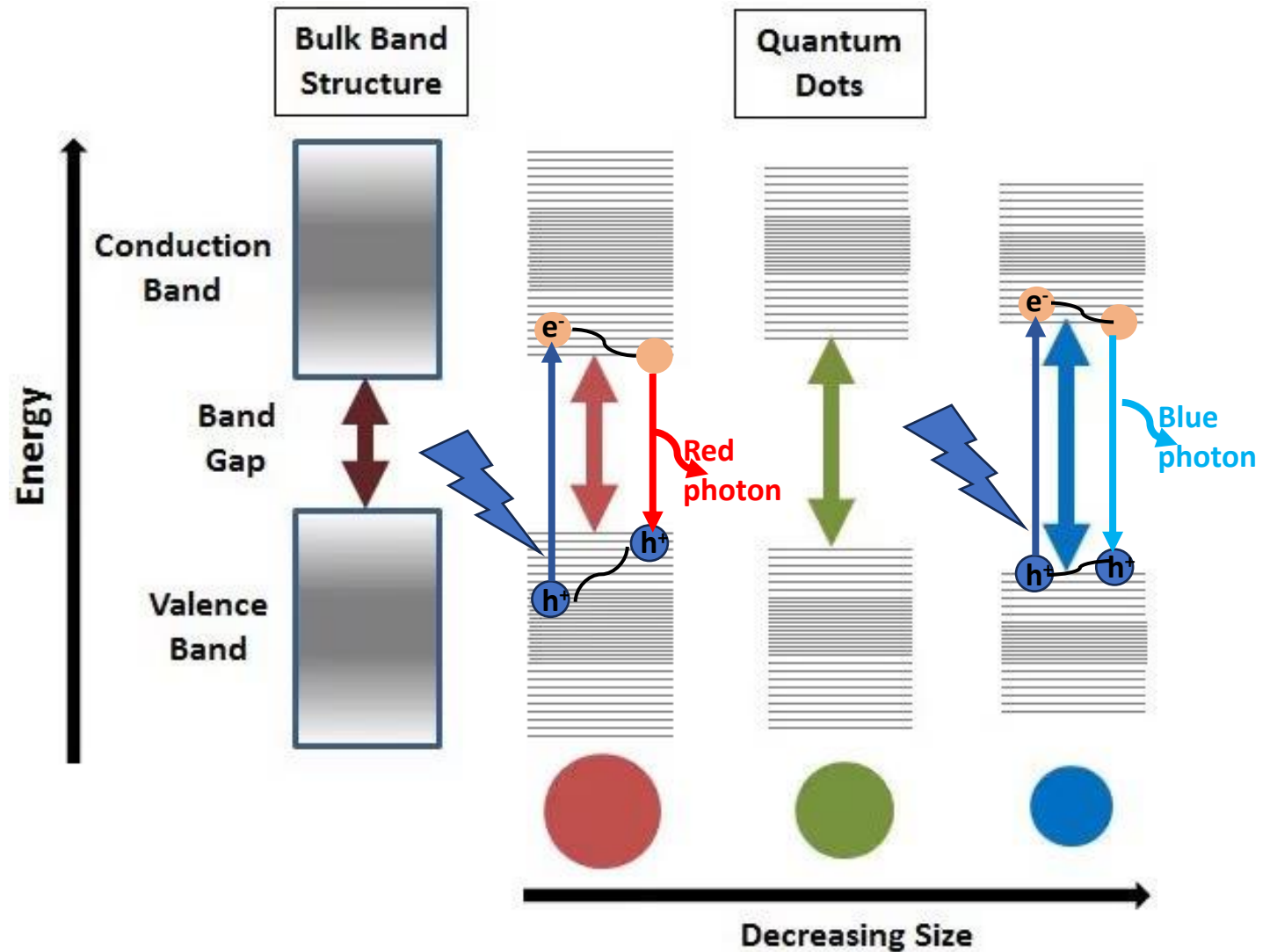
*MIT; photograph, Len Rubenstein*



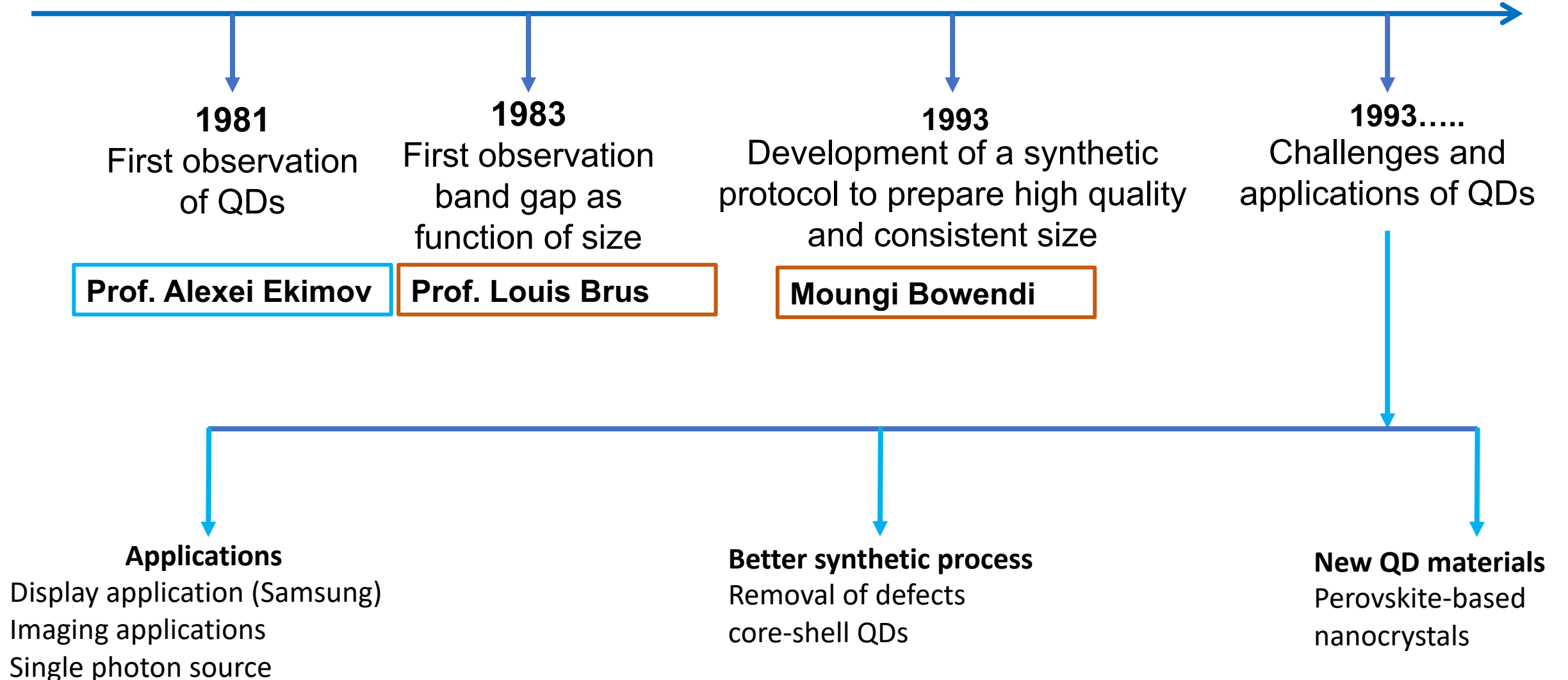
# The journey to realize size-dependent quantum effect

So, what are 'quantum dots'?

- ❑ Semiconductor nanocrystals with sizes between ~2-8 nm.
- ❑ Typically are chalcogenides (selenides, sulfides or tellurides) of metals like cadmium, lead or zinc



# The journey to realize size-dependent quantum effect



# Applications of QDs in QLED technology

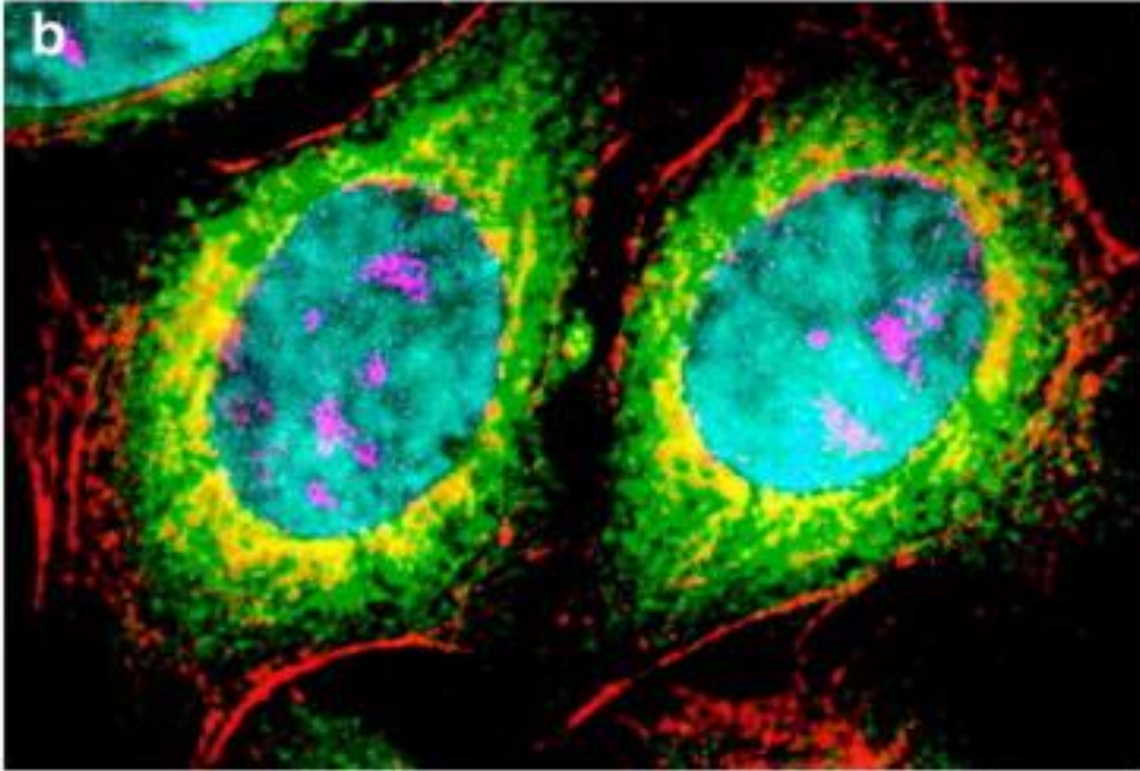
- ❖ Light-emission yield was low for typical QD materials
- ❖ Toxic materials could not be used from safety purpose
- ❖ Green and red colors are produced by QDs

‘Samsung deserves a lot of credit, they put a lot of money into the development of the technology so that they could use them in their own TVs,’ – Louis Brus.

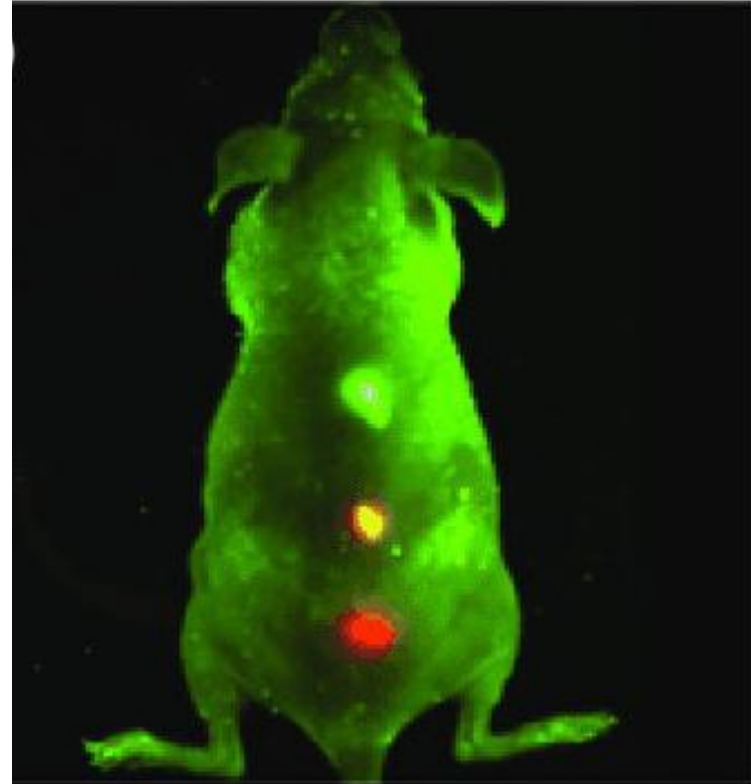
‘Cadmium selenide is not consumer friendly, even if it’s completely protected, you don’t want to be using it,’ says Bawendi. ‘[Samsung] realised that and invested a lot of money in indium phosphide as a replacement to cadmium selenide, which is a harder material to make.’



# Applications of QDs in Imaging



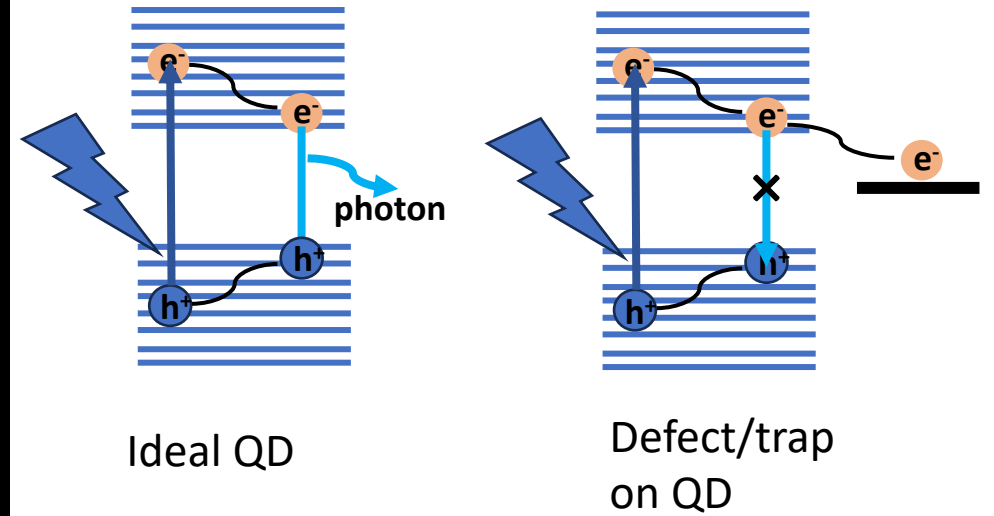
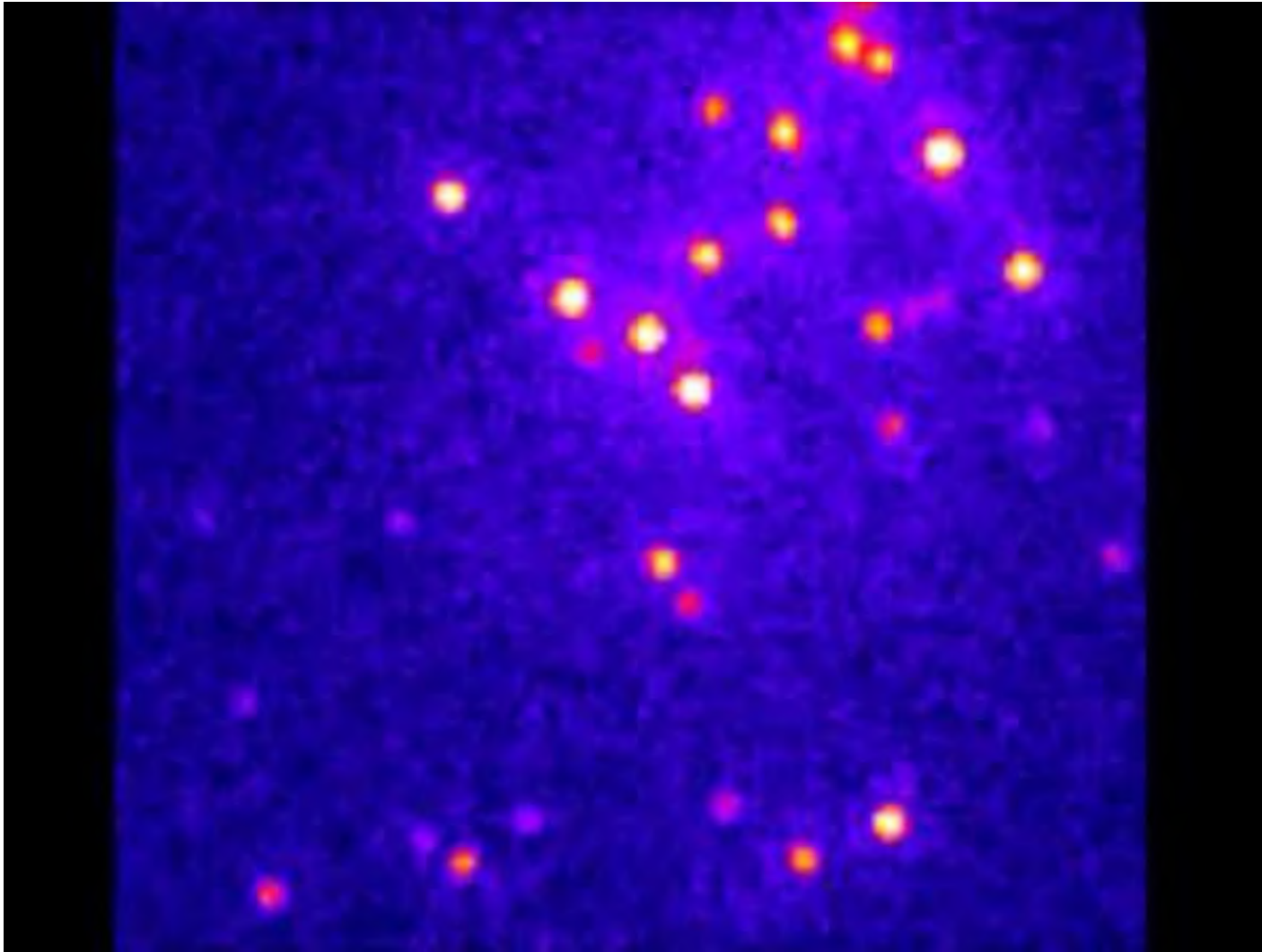
Pseudo colored image of human epithelial cells by staining with five different sizes of QDs. Actin filaments — Red (705 nm), Nucleus — Cyan (655 nm), Ki67 protein — Magenta (605 nm), Microtubules — Green (565 nm), Mitochondria — Orange (525 nm).



In vivo image of multicolor QDs injected into a live mouse

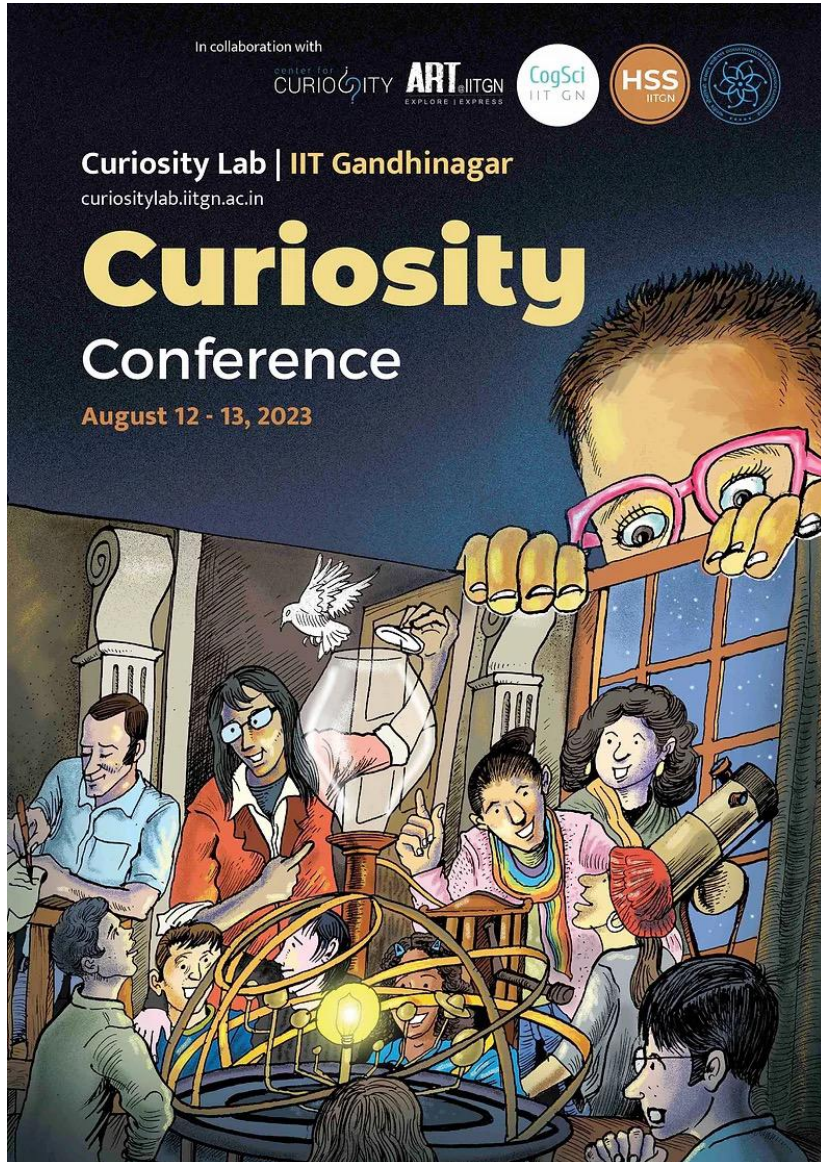


# Challenges: Defects and blinking



- ❖ Impacts light emission efficiency
- ❖ Fluctuations affect the display and imaging applications

# Conclusions and take-home message



Follow your natural curiosity



Look carefully! Your **bad data (!)** could indicate something new

Source: wikipedia



Think outside the box